

4-Phenoxy-phenol

7.34
7.30
7.28 Chloroform-d
7.07
6.98
6.95
6.84

6.12

4.75

3.80

Internal standard = 16.8 mg
 Sample = 18.6 mg
 Molar ratio = $[1/1]/[3/3] = 1$ wt% = 100.0%
 %P of standard = 1
 MW of 4-Phenoxy-phenol = 186.21
 MW of standard = 168.19

$$\text{Molar ratio} = \frac{\left[\frac{I_{cpd}}{nH_{cpd}} \right]}{\left[\frac{I_{std}}{nH_{std}} \right]}$$

$$\text{wt}\% = \frac{mg_{cpd} \times MW_{std} \times \text{molar ratio} \times P_{std}}{mg_{std} \times MW_{cpd}} \times 100$$

